ERRI/CFD Centre Faculty of Engineering



Co-Firing Coal/Biomass And The Estimation Of Burnout And NOx Formation

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Co-firing is widely used in UK power stations.

The biomass does not mill as well as coal and can lead to larger particles.

There are uncertainties about the mechanism of the formation of NO when biomass is present.

Test Coal and Co-firing Blends Studied.



- 1. Gascoigne Wood (GW) coal- used as a test coal for the CFD model
- 2. Russian coal
- 3. Russian/PKE
- 4. S. African coal
- 5. S. African coal/milled wood(20% thermal)
- 6. S. African/Miscanthus Giganteus (20% thermal)
- 7. S. African/ Olive waste (15% thermal)
- 8. S. African/Torrefied wood (20% thermal)

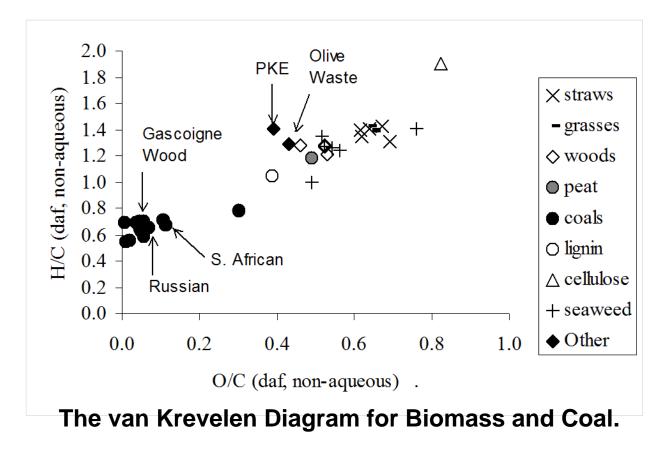
Experimental data: exit NO, T and C-in-ash, and data on radiation and flame properties for GW coal





Properties of Biomass and Coals.





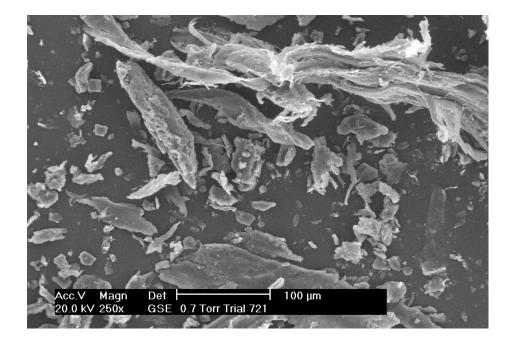
Coals and most biomass have conventional properties-PKE and Olive waste are different





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Milled Wood.



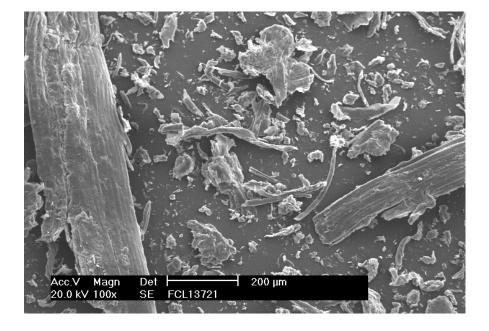
Milled Wood







Miscanthus.

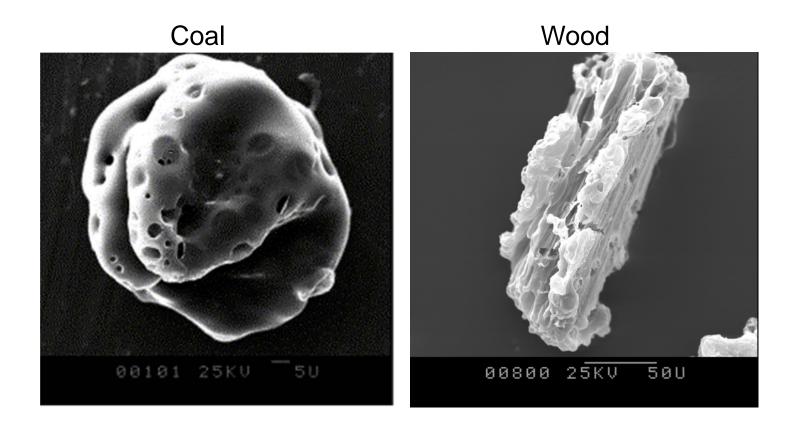


Miscanthus





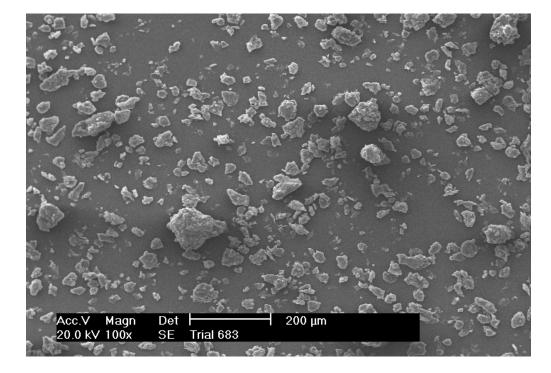
Biomass combustion: particle shape is complex.



Complex drag coefficients. Shape factor given by: f = {surface area equivalent sphere}/{actual surface area}



Palm Kernel Extractor.



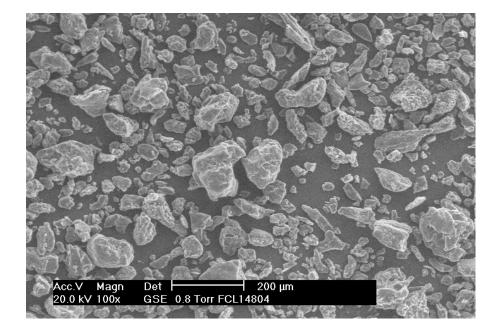
PKE







Olive Waste.



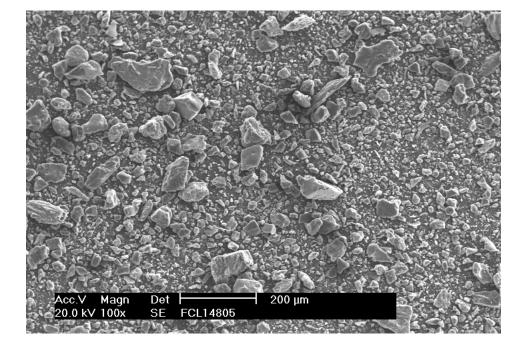
Olive Waste





South African Coal/Olive Waste.





Blend of South African coal and 15% Olive waste







Coal Reaction Mechanism.

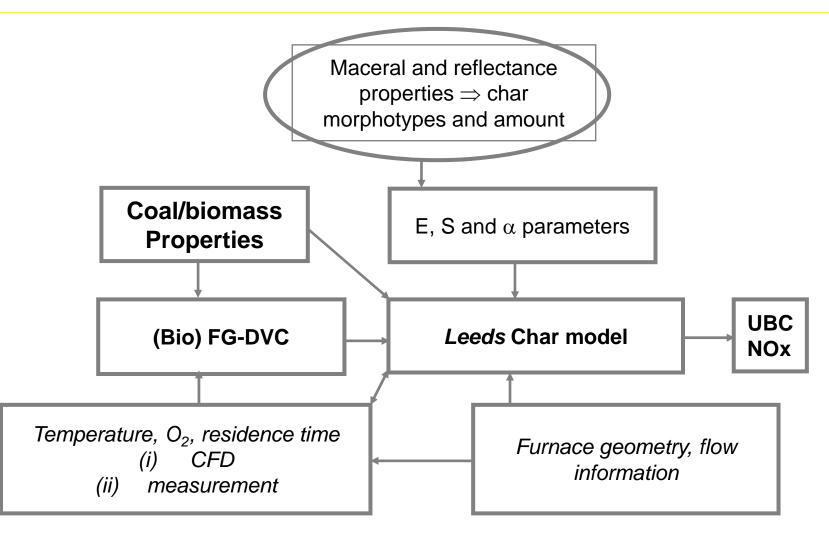
Coal = char + gases + tar	R 1
volatiles + $O_2 = CO + H_2O$	R 2
$CO+0.5 O_2 = CO_2$	R 3
C (char) + $O_2 = CO$	R 4
$CO + O_2 = CO_2$	R 5
and soot formation and burnout.	

A potassium release mechanism can be added if required.





Coal/biomass BURNOUT MODEL



Devolatilisation Rates of Coal.



Coal (Origin)	Devolatilisation Rates – FG-DVC computed						
	Activation 1	Energy, E _d (J	Pre-Exponential factor , A _d (s ⁻¹)				
	Tar	Gas	Total	Tar	Gas	Total	
Thoresby	3.3 x 10 ⁸	2.4 x 10 ⁸	2.4 x 10 ⁸	3.8 x 10 ¹⁴	4.6 x 10 ¹³	5.2 x10 ¹⁴	
Asfordby	2.0 x 10 ⁸	2.1 x 10 ⁸	2.3 x 10 ⁸	3.7 x 10 ¹⁴	4.2 x 10 ¹³	4.8 x10 ¹⁴	
Betts Lane	2.5x10 ⁸	2.3x10 ⁸	2.4x10 ⁸	2.1x10 ¹⁴	3.2x10 ¹²	3.3x10 ¹⁴	
Pittsburgh#8	2.3x10 ⁸	2.2x10 ⁸	2.3x10 ⁸	3.7x10 ¹⁴	4.8x10 ¹²	3.8x10 ¹⁴	
Ensham	2.4 x 10 ⁸	2.7 x 10 ⁸	2.4 x 10 ⁸	4.1 x10 ¹⁴	3.2 x10 ¹⁴	4.3 x10 ¹⁴	
Prodeco	2.3 x 10 ⁸	2.4 x 10 ⁸	2.4 x 10 ⁸	3.0 x10 ¹⁴	4.6 x10 ¹³	5.2 x10 ¹⁴	

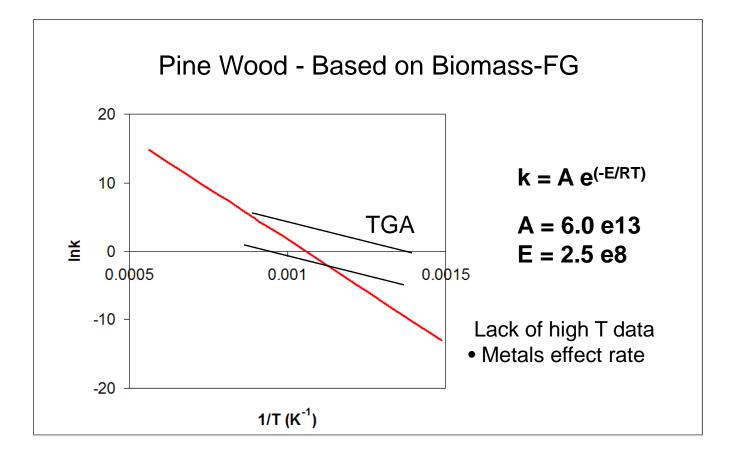
We used an averaged value for all coals.

A similarly high value was used for biomass



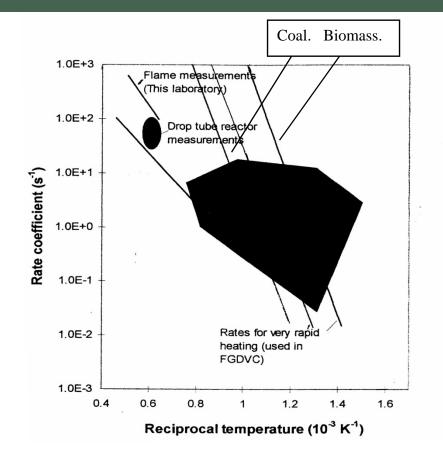


Biomass Devolatilisation Kinetics.



Kinetic Data.





Summary of Kinetic Analysis

The use of rapid rates implies that the controlling step is the heating-up process, *T*, *Cp* important.





Volatile Yields of the Coals Studied.



Volatile Yields (%wt, daf)

Coal	VM (daf)	Calculated, FG-DVC	Values from RWE npower	Experimental values: CRE, 1350º C
Gascoigne Wood coal	35.11	55.1	55.5%	48.7
Russian coal	43.43	58.1	57.5 - 69.5*	58.3
South African	37.64	54.75	48.9 - 60.2*	49.92

•From Man et al., Fuel, 2005





Volatile Yields of the Biomass Fuels.



Volatile Yields (daf)

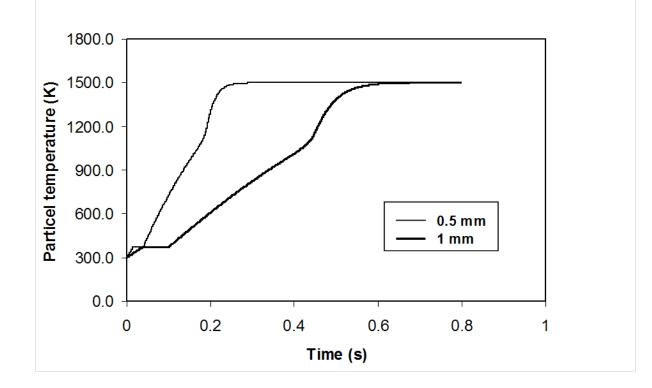
Biomass	VM (daf)	Calculated by Bio-FG	Value given by RWE npower
Milled wood	86.58.7	90.91	76.34
РКЕ	93.13	97.79	85.54
Olive	75.71	79.50	
Miscanthus	80.8	84.8	
Torrefied wood	75	80	





Heating-up Curves for Large Biomass Particles.





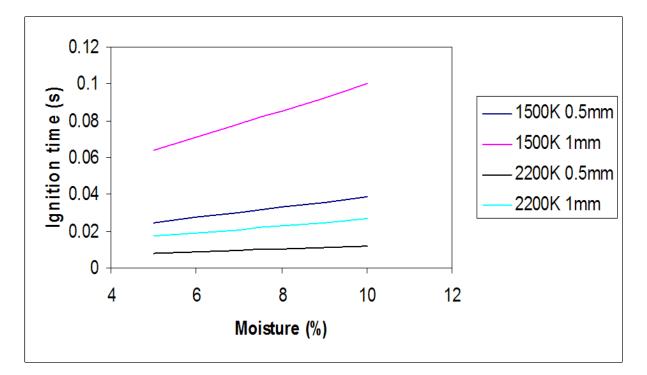
Heating-up Curves for Biomass Particles, (0.5 and 1 mm diam, 5%water content). *Cp and the water content are important*





Effect of Moisture Content on Biomass Ignition.





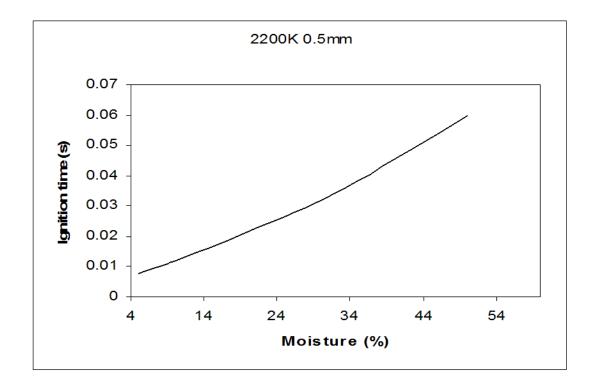
Time to Ignition (taken as the end of water evaporation) as a function of moisture content for different sized particles and temperature.





Effect of High Moisture Content.





Ignition times for high moisture levels, 0.5 mm and 2200K.







The formation of soot was estimated by the equation below-

$$\mathcal{R}_{\rm soot, form} = C_s p_{\rm fuel} \phi^r e^{-E/RT}$$

The combustion of the soot particles was governed by the Magnussen equation.



RWE npower CTF

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Modelling the RWE npower CTF.





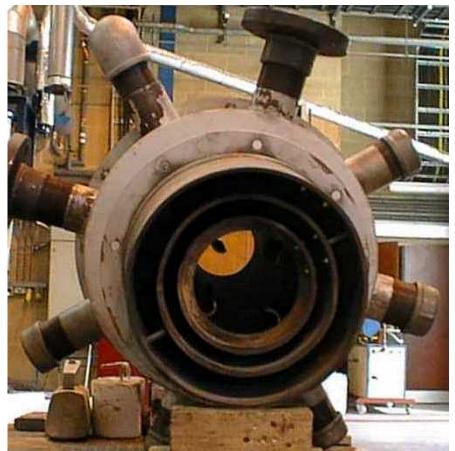
Furnace Geometry





Mitsui Babcock Burner.





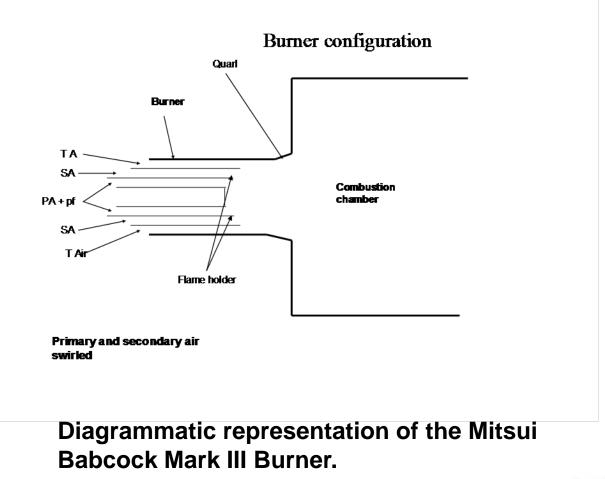
Burner Face Details.





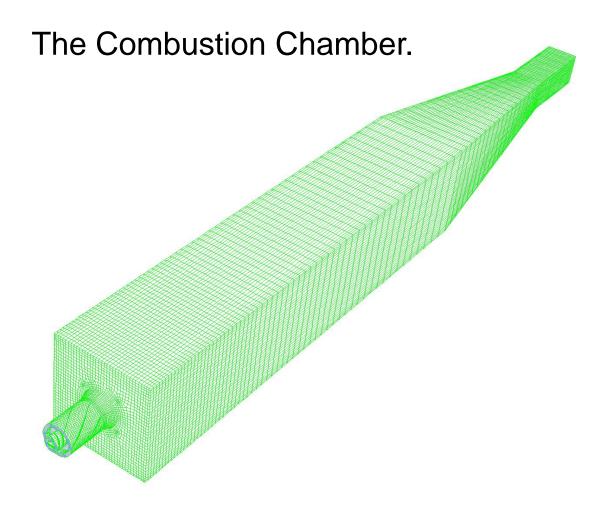
Burner Details.





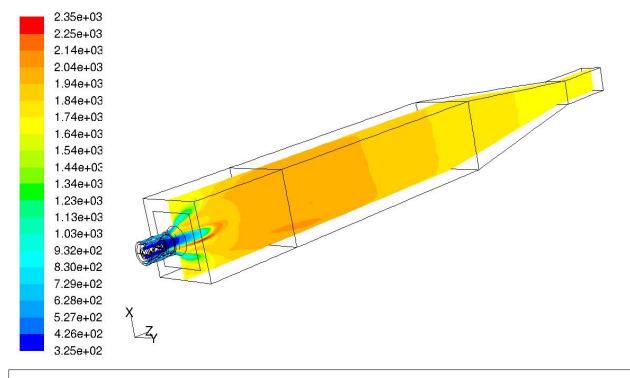






Typical Computed Temperatures.





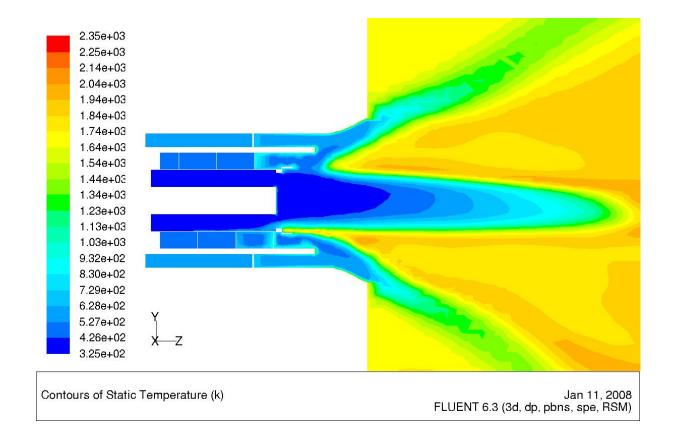
Contours of Static Temperature (k) Dec 21, 2007 FLUENT 6.3 (3d, dp, pbns, spe, RSM)





Computation of the Near Burner Region.







Photograph of the Flame.





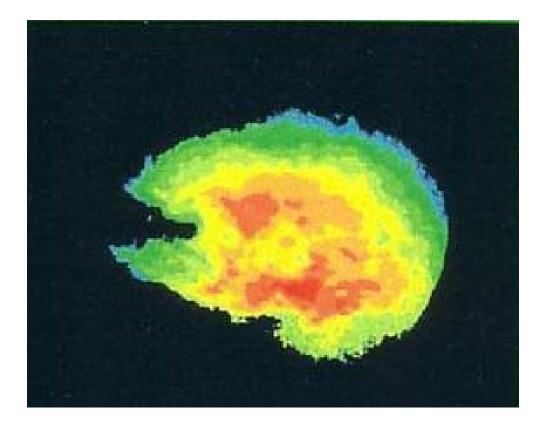
Photograph of the Flame





Measured Temperature Field.





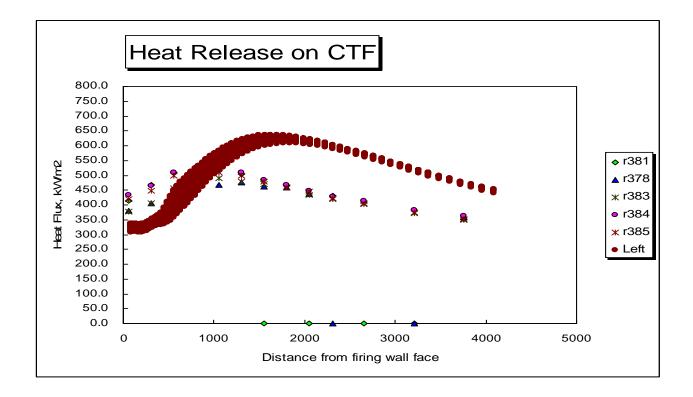
Measured Temperature Field (Professor Yan): red 1650°C; yellow,1580°C; blue/green,1450°C.





Measured and Computed Incident Radiation.





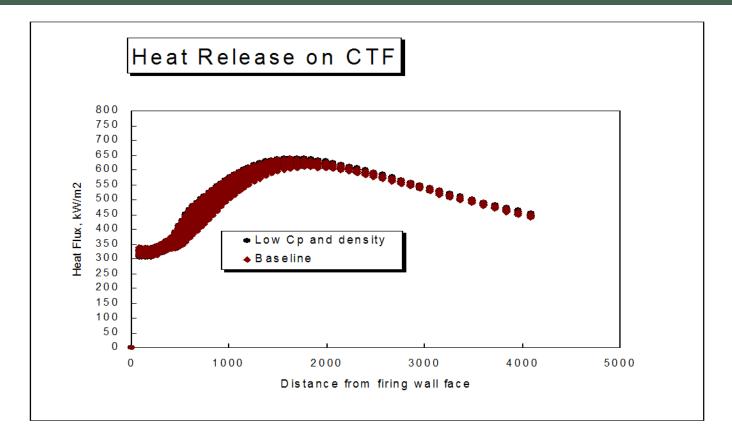
Surface Incident Radiation Measurements and Computed Values for Gascoigne Wood Coal.





Measured and Computed Incident Radiation-Effect of Cp and Density.





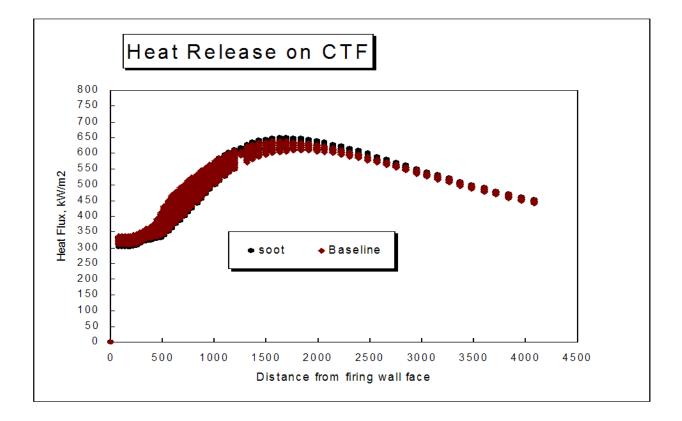
Surface Incident Radiation. Effect of Lowering Cp and Density of Fuel Particles (Gascoigne Wood) (both lowered about 10%).





Effect of Changing Rate of Soot Formation.





Surface Incident Radiation. Effect of Soot Formation is small (Gascoigne Wood).



Co-firing - Irregular Shaped Biomass Particles.



Aerodynamic Factors

The drag coefficients, C_D , were calculated as below and shape factors, f, were used to account for the irregular biomass particles shapes.

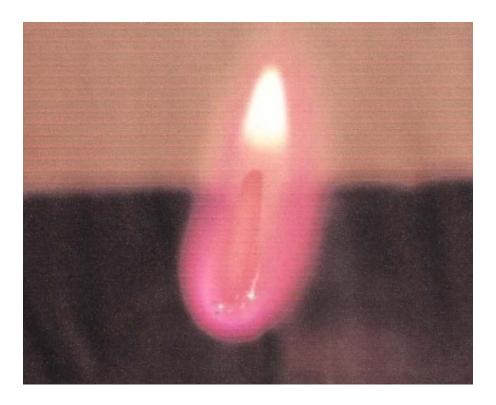
$$C_{D} = (24/Re) (1 + b_1Re^{b^2}) + [b_3 Re/(b_4 + Re)]$$

 $b_1 = 2.3288 - 6.4581f + 2.4486f^2$ $b_2 = 0.0964 + 0.5565f$ $b_3 = 4.9050 - 13.8944f + 18.4222f^2 - 10.2599f^3$ $b_4 = 1.4681 + 12.258f - 20.7322f^2 + 15.8855f^3$



Examination of Combustion of Single Particles of Biomass.





Examination of the Combustion of a Single Particle of Wood

There is little disruptive burning. A major issue is the use of equivalent spheres in the combustion model







The NO model used in Fluent permits the N-component in the fuel to be converted to HCN or NH_3 which is then oxidised to NO and N_2 depending on the reaction conditions.

In coal it is known that the fuel-N converts to HCN. In biomass the fuel–N is in the form of proteins or amino acids and is converted to a mixture of $NH_3 + HCN$.

For the biomass used here we have allowed it to be converted to 90% NH_3 / 10% HCN which then reacts to form NO.



Co-firing Biomass and Coal.





Experimental Furnace Flame Shape. The flame divergence is greater with biomass



Computational Fluid Dynamics

Experimental and Computed Results for Single Coals and Co-firing Blends ..



Fuel	Calc NO ppm (dry)	Measured NO ppm (dry)	Calc C in Ash, Mass %	Measured C in Ash, Mass %	Calculated exit T, K	Experimental Exit T, K
Gascoigne Wood coal	300	325	1.32	3.0	1605	1597
Russian coal	296	325	1.24	3.4	1617	1407
Russian/PKE	316	321	3.5	4.8	1487	1488
S. African coal	292	312	5.87	6.7	1668	1424
S. African coal/ wood	288	319	5.6	5.9	1583	1527
S. African/ Miscanthus (20%)	250	242	8.9	1.94	1664	1470
S. African/ Olive waste (15%)	276	230	0.62	0.98	1646	1539
S. African/Torrefied Wood* (20% thermal)	264	Not available	0.3	Not available	1724	Not available

Comparison of Experimental and Computed Data.





Conclusions.



Conclusions

•A coal combustion model has been tested using Gascoigne Wood coal with the RWE npower Combustion Test Facility. Reasonable agreement has been obtained for exit temperatures, NO, unburned carbon-in-ash. However agreement with the radiation measurements is not good. The influence of heating-up of the particles, and the devolatilisation and soot models have been examined, but the key factor seems to be the correct aerodynamic modelling of the burner.

•In the case of biomass the behaviour of water content has been examined and the effect on the release of volatiles from large biomass particles examined.

•Co-firing of wood, miscanthus, PKE and olive wastes with a number of coals has been modelled with reasonable success..



